

## 2,3,4,6-Tetra-O-acetyl- $\beta$ -D-galactopyranosyl 2-(2,4-dichloroanilino)-4,4-dimethyl-6-oxocyclohex-1-enecarbo-dithioate

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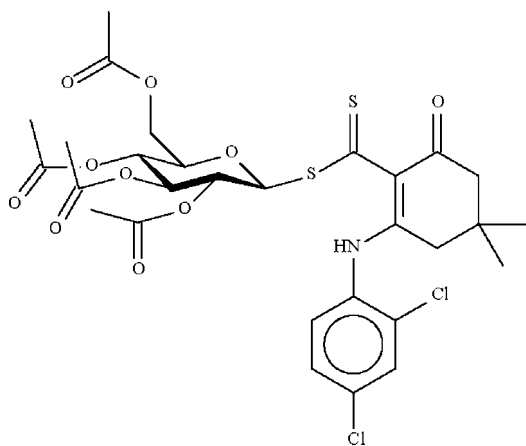
Received 17 April 2009; accepted 21 April 2009

Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.011$  Å;  $R$  factor = 0.092;  $wR$  factor = 0.263; data-to-parameter ratio = 18.1.

The cyclohexene ring in the title compound,  $\text{C}_{29}\text{H}_{33}\text{Cl}_2\text{NO}_{10}\text{S}_2$ , adopts an envelope conformation, with the C atom bearing the two methyl groups representing the flap. This atom deviates by 0.63 (1) Å from the plane through the other five ring atoms (r.m.s. deviation = 0.11 Å). The molecular conformation is stabilized by an intramolecular  $\text{N}-\text{H}\cdots\text{S}$  hydrogen bond. The crystal studied was a non-merohedral twin, with a minor twin component of 29%.

### Related literature

For background to thioglycosides, see: El Ashry *et al.* (2006, 2008), Haikel *et al.* (2003). For the deconvolution of non-merohedrally twinned diffraction data, see: Spek (2009).



### Experimental

#### Crystal data

$\text{C}_{29}\text{H}_{33}\text{Cl}_2\text{NO}_{10}\text{S}_2$   
 $M_r = 690.58$   
 Monoclinic,  $P2_1$   
 $a = 13.8257$  (4) Å  
 $b = 8.7697$  (3) Å  
 $c = 14.0690$  (4) Å  
 $\beta = 106.486$  (2)°

$V = 1635.70$  (9) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.38$  mm<sup>-1</sup>  
 $T = 100$  K  
 $0.35 \times 0.15 \times 0.02$  mm

#### Data collection

Bruker SMART APEX diffractometer  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.852$ ,  $T_{\max} = 0.992$

15059 measured reflections  
 7402 independent reflections  
 5732 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.065$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.092$   
 $wR(F^2) = 0.263$   
 $S = 1.09$   
 7402 reflections  
 408 parameters  
 2 restraints

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 1.06$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.99$  e Å<sup>-3</sup>  
 Absolute structure: Flack (1983), 3420 Friedel pairs  
 Flack parameter: 0.1 (2)

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$                 | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--------------------------------------|--------------|--------------------|-------------|----------------------|
| $\text{N1}-\text{H1}\cdots\text{S2}$ | 0.88 (1)     | 2.07 (5)           | 2.882 (6)   | 152 (9)              |

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2009).

We thank the Higher Education Commission of Pakistan and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK2430).

### References

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 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.  
 Westrip, S. P. (2009). publCIF. In preparation.

**supplementary materials**

*Acta Cryst.* (2009). E65, o1106 [ doi:10.1107/S1600536809014743 ]

**2,3,4,6-Tetra-*O*-acetyl- $\beta$ -D-galactopyranosyl 2-(2,4-dichloroanilino)-4,4-dimethyl-6-oxocyclohex-1-enecarbodithioate**

**E. S. H. El Ashry, M. R. Amer, M. R. Shah and S. W. Ng**

**Experimental**

A cooled (283 K) solution of (2,4-dichloroanilino)-5,5-dimethyl-cyclohex-2-en-1-one (0.1 mol) and sodium hydroxide (0.4 g) in DMSO (20 ml) and water (1 ml) was treated with carbon disulfide (0.3 mol). After 20 min, 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-galactopyranosyl bromide (0.12 mol) was added and the reaction mixture was left for 24 h. Water (200 ml) was added and the mixture acidified with 10% hydrochloric acid. The product was purified on by silica-gel column-chromatography to give yellow crystals that were further crystallized from methanol (m.p. 480 K).

**Refinement**

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95 to 1.00 Å) and were included in the refinement in the riding model approximation, with  $U(H)$  set to 1.2 to 1.5 $U(C)$ . The amino H-atom was located in a difference Fourier map and was refined with a distance restraint of N—H 0.88±0.01 Å; its displacement factor was freely refined.

The structure initially refined to  $R = 0.118$  but the displacement factors and bond dimensions for all atoms were normal. Subsequent analysis showed the structure to be a non-merohedral twin. *PLATON* (Spek, 2009) split the reflection data by the matrix (-1 0 0, 0 - 1 0, 0.578 0 1). The minor twin component refined to 0.287.

The final difference Fourier map had a large peak at 1.5 Å from H4a.

**Figures**

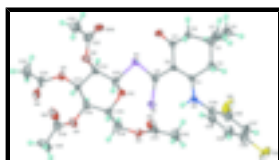


Fig. 1. Thermal ellipsoid plot (Barbour, 2001) plot of  $C_{29}H_{33}Cl_2NO_{10}S_2$  at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

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*Crystal data*

$C_{29}H_{33}Cl_2NO_{10}S_2$

$M_r = 690.58$

Monoclinic,  $P2_1$

Hall symbol: P 2yb

$a = 13.8257(4)$  Å

$F_{000} = 720$

$D_x = 1.402$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 2620 reflections

$\theta = 2.8\text{--}21.7^\circ$

# supplementary materials

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$b = 8.7697 (3) \text{ \AA}$   
 $c = 14.0690 (4) \text{ \AA}$   
 $\beta = 106.486 (2)^\circ$   
 $V = 1635.70 (9) \text{ \AA}^3$   
 $Z = 2$

$\mu = 0.38 \text{ mm}^{-1}$   
 $T = 100 \text{ K}$   
Plate, orange  
 $0.35 \times 0.15 \times 0.02 \text{ mm}$

## Data collection

Bruker SMART APEX diffractometer  
Radiation source: fine-focus sealed tube  
Monochromator: graphite  
 $T = 100 \text{ K}$   
 $\omega$  scans  
Absorption correction: Multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.852$ ,  $T_{\max} = 0.992$   
15059 measured reflections

7402 independent reflections  
5732 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.065$   
 $\theta_{\max} = 27.5^\circ$   
 $\theta_{\min} = 1.8^\circ$   
 $h = -17 \rightarrow 17$   
 $k = -11 \rightarrow 11$   
 $l = -18 \rightarrow 18$

## Refinement

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.092$   
 $wR(F^2) = 0.263$   
 $S = 1.09$   
7402 reflections  
408 parameters  
2 restraints  
Primary atom site location: structure-invariant direct methods  
Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites  
H atoms treated by a mixture of independent and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.1355P)^2 + 3.0358P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 1.06 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.99 \text{ e \AA}^{-3}$   
Extinction correction: none  
Absolute structure: Flack (1983), 3420 Friedel pairs  
Flack parameter: 0.1 (2)

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|     | x            | y           | z             | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|-------------|---------------|----------------------------------|
| S1  | 0.78612 (13) | 0.4998 (2)  | 0.07646 (13)  | 0.0206 (4)                       |
| S2  | 0.70294 (13) | 0.2505 (2)  | -0.06603 (14) | 0.0213 (4)                       |
| Cl1 | 0.7299 (2)   | 0.4115 (3)  | -0.39454 (16) | 0.0473 (6)                       |
| Cl2 | 0.8139 (2)   | -0.0943 (3) | -0.57216 (16) | 0.0449 (6)                       |
| O1  | 0.9601 (5)   | 0.5765 (7)  | 0.0630 (4)    | 0.0350 (14)                      |
| O2  | 0.5862 (4)   | 0.4620 (5)  | 0.0353 (4)    | 0.0192 (10)                      |
| O3  | 0.4399 (4)   | 0.3463 (6)  | -0.1291 (4)   | 0.0214 (11)                      |
| O4  | 0.3754 (5)   | 0.5700 (6)  | -0.1994 (4)   | 0.0327 (13)                      |
| O5  | 0.4186 (4)   | 0.2941 (6)  | 0.1614 (4)    | 0.0238 (11)                      |

|      |            |             |             |             |
|------|------------|-------------|-------------|-------------|
| O6   | 0.3331 (5) | 0.4616 (7)  | 0.2260 (5)  | 0.0368 (15) |
| O7   | 0.5631 (4) | 0.5043 (6)  | 0.3097 (3)  | 0.0216 (10) |
| O8   | 0.5287 (5) | 0.2970 (7)  | 0.3894 (4)  | 0.0328 (14) |
| O9   | 0.7568 (4) | 0.4371 (6)  | 0.2862 (4)  | 0.0245 (11) |
| O10  | 0.8313 (5) | 0.6649 (7)  | 0.2879 (4)  | 0.0380 (15) |
| N1   | 0.8206 (4) | 0.2397 (8)  | -0.2053 (4) | 0.0204 (12) |
| H1   | 0.774 (5)  | 0.218 (11)  | -0.176 (6)  | 0.03 (3)*   |
| C1   | 0.8882 (5) | 0.3347 (8)  | -0.1512 (5) | 0.0167 (13) |
| C2   | 0.9760 (5) | 0.3714 (8)  | -0.1929 (5) | 0.0211 (15) |
| H2A  | 1.0162     | 0.2773      | -0.1904     | 0.025*      |
| H2B  | 0.9479     | 0.3992      | -0.2636     | 0.025*      |
| C3   | 1.0468 (5) | 0.4979 (9)  | -0.1419 (6) | 0.0231 (14) |
| C4   | 1.0629 (5) | 0.4783 (9)  | -0.0316 (5) | 0.0209 (15) |
| H4A  | 1.0920     | 0.3764      | -0.0108     | 0.025*      |
| H4B  | 1.1113     | 0.5560      | 0.0048      | 0.025*      |
| C5   | 0.9649 (5) | 0.4947 (9)  | -0.0061 (5) | 0.0224 (15) |
| C6   | 0.8785 (5) | 0.4039 (9)  | -0.0634 (5) | 0.0199 (14) |
| C7   | 1.0011 (7) | 0.6559 (9)  | -0.1783 (7) | 0.0316 (18) |
| H7A  | 0.9774     | 0.6565      | -0.2509     | 0.047*      |
| H7B  | 1.0527     | 0.7348      | -0.1556     | 0.047*      |
| H7C  | 0.9443     | 0.6764      | -0.1514     | 0.047*      |
| C8   | 1.1460 (6) | 0.4828 (11) | -0.1689 (7) | 0.0336 (19) |
| H8A  | 1.1726     | 0.3792      | -0.1537     | 0.050*      |
| H8B  | 1.1949     | 0.5565      | -0.1304     | 0.050*      |
| H8C  | 1.1341     | 0.5029      | -0.2398     | 0.050*      |
| C9   | 0.7936 (5) | 0.3796 (8)  | -0.0239 (5) | 0.0189 (14) |
| C11  | 0.6779 (5) | 0.4207 (8)  | 0.1073 (5)  | 0.0181 (13) |
| H11  | 0.6838     | 0.3071      | 0.1115      | 0.022*      |
| C12  | 0.6713 (5) | 0.4842 (9)  | 0.2059 (5)  | 0.0208 (14) |
| H12  | 0.6652     | 0.5979      | 0.2031      | 0.025*      |
| C13  | 0.5805 (5) | 0.4115 (8)  | 0.2314 (5)  | 0.0171 (13) |
| H13  | 0.5976     | 0.3051      | 0.2559      | 0.021*      |
| C14  | 0.4841 (5) | 0.4106 (9)  | 0.1444 (5)  | 0.0207 (14) |
| H14  | 0.4501     | 0.5123      | 0.1375      | 0.025*      |
| C15  | 0.5079 (5) | 0.3665 (8)  | 0.0481 (5)  | 0.0185 (14) |
| H15  | 0.5297     | 0.2574      | 0.0513      | 0.022*      |
| C16  | 0.4140 (6) | 0.3897 (9)  | -0.0387 (5) | 0.0252 (16) |
| H16A | 0.3581     | 0.3254      | -0.0304     | 0.030*      |
| H16B | 0.3924     | 0.4977      | -0.0425     | 0.030*      |
| C17  | 0.4192 (6) | 0.4505 (9)  | -0.2029 (6) | 0.0245 (16) |
| C18  | 0.4605 (7) | 0.3996 (10) | -0.2836 (6) | 0.0339 (18) |
| H18A | 0.5175     | 0.4648      | -0.2855     | 0.051*      |
| H18B | 0.4079     | 0.4066      | -0.3471     | 0.051*      |
| H18C | 0.4834     | 0.2937      | -0.2719     | 0.051*      |
| C19  | 0.3470 (6) | 0.3317 (10) | 0.2036 (6)  | 0.0281 (17) |
| C20  | 0.2888 (8) | 0.1967 (12) | 0.2189 (9)  | 0.052 (3)   |
| H20A | 0.2653     | 0.2126      | 0.2777      | 0.077*      |
| H20B | 0.3319     | 0.1061      | 0.2285      | 0.077*      |
| H20C | 0.2306     | 0.1823      | 0.1607      | 0.077*      |

## supplementary materials

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|      |            |             |             |             |
|------|------------|-------------|-------------|-------------|
| C21  | 0.5345 (6) | 0.4330 (9)  | 0.3822 (6)  | 0.0252 (16) |
| C22  | 0.5020 (7) | 0.5445 (11) | 0.4462 (7)  | 0.037 (2)   |
| H22A | 0.5078     | 0.4982      | 0.5110      | 0.056*      |
| H22B | 0.4318     | 0.5738      | 0.4150      | 0.056*      |
| H22C | 0.5451     | 0.6352      | 0.4550      | 0.056*      |
| C23  | 0.8296 (6) | 0.5446 (11) | 0.3232 (7)  | 0.035 (2)   |
| C24  | 0.8238 (5) | 0.1667 (9)  | -0.2938 (5) | 0.0215 (15) |
| C25  | 0.9019 (7) | 0.4892 (15) | 0.4157 (7)  | 0.050 (3)   |
| H25A | 0.9698     | 0.5260      | 0.4192      | 0.075*      |
| H25B | 0.9018     | 0.3774      | 0.4164      | 0.075*      |
| H25C | 0.8821     | 0.5275      | 0.4729      | 0.075*      |
| C26  | 0.7794 (6) | 0.2291 (9)  | -0.3864 (6) | 0.0257 (16) |
| C27  | 0.7763 (7) | 0.1509 (11) | -0.4737 (6) | 0.0301 (17) |
| H27  | 0.7479     | 0.1967      | -0.5367     | 0.036*      |
| C28  | 0.8154 (7) | 0.0059 (11) | -0.4659 (6) | 0.0321 (18) |
| C29  | 0.8613 (7) | -0.0591 (9) | -0.3748 (6) | 0.0320 (19) |
| H29  | 0.8900     | -0.1581     | -0.3712     | 0.038*      |
| C30  | 0.8655 (6) | 0.0206 (9)  | -0.2893 (6) | 0.0253 (16) |
| H30  | 0.8970     | -0.0242     | -0.2266     | 0.030*      |

### Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$    | $U^{22}$    | $U^{33}$    | $U^{12}$    | $U^{13}$     | $U^{23}$     |
|-----|-------------|-------------|-------------|-------------|--------------|--------------|
| S1  | 0.0217 (8)  | 0.0199 (8)  | 0.0235 (8)  | -0.0054 (7) | 0.0115 (7)   | -0.0049 (7)  |
| S2  | 0.0182 (8)  | 0.0227 (8)  | 0.0256 (9)  | -0.0056 (7) | 0.0107 (7)   | -0.0072 (7)  |
| C11 | 0.0630 (15) | 0.0380 (12) | 0.0311 (11) | 0.0245 (12) | -0.0025 (10) | -0.0023 (9)  |
| C12 | 0.0575 (14) | 0.0491 (14) | 0.0272 (10) | 0.0067 (12) | 0.0103 (10)  | -0.0145 (10) |
| O1  | 0.037 (3)   | 0.045 (3)   | 0.028 (3)   | -0.023 (3)  | 0.017 (3)    | -0.019 (3)   |
| O2  | 0.018 (2)   | 0.019 (2)   | 0.019 (2)   | 0.0023 (19) | 0.004 (2)    | 0.0006 (18)  |
| O3  | 0.027 (3)   | 0.022 (3)   | 0.016 (2)   | 0.001 (2)   | 0.007 (2)    | -0.0001 (19) |
| O4  | 0.044 (4)   | 0.021 (3)   | 0.038 (3)   | 0.003 (3)   | 0.019 (3)    | 0.002 (2)    |
| O5  | 0.017 (2)   | 0.026 (3)   | 0.032 (3)   | -0.001 (2)  | 0.014 (2)    | 0.000 (2)    |
| O6  | 0.034 (3)   | 0.029 (3)   | 0.056 (4)   | 0.009 (2)   | 0.028 (3)    | 0.011 (3)    |
| O7  | 0.027 (3)   | 0.023 (2)   | 0.017 (2)   | 0.005 (2)   | 0.011 (2)    | -0.001 (2)   |
| O8  | 0.048 (4)   | 0.031 (3)   | 0.024 (3)   | -0.005 (3)  | 0.016 (3)    | 0.002 (2)    |
| O9  | 0.019 (3)   | 0.027 (3)   | 0.026 (3)   | -0.004 (2)  | 0.004 (2)    | -0.002 (2)   |
| O10 | 0.051 (4)   | 0.035 (3)   | 0.030 (3)   | -0.022 (3)  | 0.014 (3)    | -0.014 (3)   |
| N1  | 0.016 (3)   | 0.028 (3)   | 0.020 (3)   | -0.005 (3)  | 0.010 (2)    | -0.003 (2)   |
| C1  | 0.014 (3)   | 0.016 (3)   | 0.019 (3)   | -0.002 (3)  | 0.003 (3)    | -0.001 (3)   |
| C2  | 0.020 (3)   | 0.023 (4)   | 0.022 (3)   | 0.003 (3)   | 0.008 (3)    | -0.001 (3)   |
| C3  | 0.016 (3)   | 0.026 (4)   | 0.027 (4)   | -0.006 (3)  | 0.005 (3)    | 0.000 (3)    |
| C4  | 0.013 (3)   | 0.028 (4)   | 0.021 (3)   | -0.007 (3)  | 0.005 (3)    | -0.003 (3)   |
| C5  | 0.025 (4)   | 0.025 (3)   | 0.023 (3)   | -0.006 (3)  | 0.016 (3)    | -0.007 (3)   |
| C6  | 0.018 (3)   | 0.024 (4)   | 0.019 (3)   | -0.008 (3)  | 0.008 (3)    | -0.002 (3)   |
| C7  | 0.038 (5)   | 0.022 (4)   | 0.038 (5)   | 0.000 (4)   | 0.016 (4)    | -0.001 (3)   |
| C8  | 0.029 (4)   | 0.040 (5)   | 0.039 (5)   | -0.010 (4)  | 0.022 (4)    | -0.013 (4)   |
| C9  | 0.018 (3)   | 0.017 (3)   | 0.020 (3)   | 0.003 (3)   | 0.004 (3)    | 0.004 (3)    |
| C11 | 0.016 (3)   | 0.021 (3)   | 0.016 (3)   | 0.005 (3)   | 0.002 (3)    | 0.002 (3)    |

|     |           |           |           |            |           |            |
|-----|-----------|-----------|-----------|------------|-----------|------------|
| C12 | 0.022 (3) | 0.022 (4) | 0.021 (3) | 0.000 (3)  | 0.011 (3) | 0.003 (3)  |
| C13 | 0.018 (3) | 0.019 (3) | 0.015 (3) | -0.002 (3) | 0.005 (3) | -0.004 (3) |
| C14 | 0.023 (4) | 0.019 (3) | 0.025 (4) | 0.002 (3)  | 0.015 (3) | 0.001 (3)  |
| C15 | 0.010 (3) | 0.025 (4) | 0.020 (3) | 0.002 (3)  | 0.003 (3) | -0.005 (3) |
| C16 | 0.025 (4) | 0.029 (4) | 0.019 (3) | 0.006 (3)  | 0.003 (3) | 0.000 (3)  |
| C17 | 0.025 (4) | 0.024 (4) | 0.024 (4) | -0.004 (3) | 0.007 (3) | -0.003 (3) |
| C18 | 0.040 (5) | 0.032 (4) | 0.033 (4) | 0.003 (4)  | 0.015 (4) | 0.008 (4)  |
| C19 | 0.020 (4) | 0.038 (5) | 0.028 (4) | 0.004 (3)  | 0.011 (3) | 0.007 (3)  |
| C20 | 0.030 (5) | 0.048 (6) | 0.085 (8) | -0.013 (4) | 0.030 (5) | -0.003 (6) |
| C21 | 0.024 (4) | 0.031 (4) | 0.023 (4) | 0.002 (3)  | 0.011 (3) | 0.005 (3)  |
| C22 | 0.045 (5) | 0.041 (5) | 0.037 (5) | 0.010 (4)  | 0.030 (4) | 0.000 (4)  |
| C23 | 0.029 (4) | 0.046 (6) | 0.033 (5) | -0.006 (4) | 0.014 (4) | -0.022 (4) |
| C24 | 0.018 (3) | 0.028 (4) | 0.018 (3) | -0.003 (3) | 0.006 (3) | -0.005 (3) |
| C25 | 0.037 (5) | 0.074 (7) | 0.034 (5) | -0.025 (5) | 0.002 (4) | 0.000 (5)  |
| C26 | 0.022 (4) | 0.030 (4) | 0.025 (4) | 0.005 (3)  | 0.006 (3) | -0.004 (3) |
| C27 | 0.032 (4) | 0.039 (4) | 0.017 (4) | 0.002 (4)  | 0.004 (3) | -0.008 (3) |
| C28 | 0.036 (4) | 0.038 (5) | 0.024 (4) | 0.001 (4)  | 0.011 (3) | -0.006 (4) |
| C29 | 0.046 (5) | 0.023 (4) | 0.029 (4) | 0.017 (4)  | 0.015 (4) | 0.003 (3)  |
| C30 | 0.035 (4) | 0.022 (4) | 0.023 (4) | 0.005 (3)  | 0.013 (3) | 0.004 (3)  |

*Geometric parameters (Å, °)*

|         |            |          |            |
|---------|------------|----------|------------|
| S1—C9   | 1.788 (7)  | C8—H8B   | 0.9800     |
| S1—C11  | 1.811 (7)  | C8—H8C   | 0.9800     |
| S2—C9   | 1.668 (7)  | C11—C12  | 1.521 (9)  |
| C11—C26 | 1.730 (8)  | C11—H11  | 1.0000     |
| C12—C28 | 1.729 (8)  | C12—C13  | 1.539 (9)  |
| O1—C5   | 1.226 (9)  | C12—H12  | 1.0000     |
| O2—C15  | 1.420 (8)  | C13—C14  | 1.532 (10) |
| O2—C11  | 1.426 (8)  | C13—H13  | 1.0000     |
| O3—C17  | 1.351 (9)  | C14—C15  | 1.532 (9)  |
| O3—C16  | 1.466 (9)  | C14—H14  | 1.0000     |
| O4—C17  | 1.219 (10) | C15—C16  | 1.523 (9)  |
| O5—C19  | 1.332 (9)  | C15—H15  | 1.0000     |
| O5—C14  | 1.429 (9)  | C16—H16A | 0.9900     |
| O6—C19  | 1.212 (10) | C16—H16B | 0.9900     |
| O7—C21  | 1.348 (9)  | C17—C18  | 1.477 (11) |
| O7—C13  | 1.444 (8)  | C18—H18A | 0.9800     |
| O8—C21  | 1.201 (10) | C18—H18B | 0.9800     |
| O9—C23  | 1.369 (10) | C18—H18C | 0.9800     |
| O9—C12  | 1.445 (9)  | C19—C20  | 1.481 (12) |
| O10—C23 | 1.169 (12) | C20—H20A | 0.9800     |
| N1—C1   | 1.321 (9)  | C20—H20B | 0.9800     |
| N1—C24  | 1.411 (9)  | C20—H20C | 0.9800     |
| N1—H1   | 0.881 (10) | C21—C22  | 1.483 (11) |
| C1—C6   | 1.415 (10) | C22—H22A | 0.9800     |
| C1—C2   | 1.525 (10) | C22—H22B | 0.9800     |
| C2—C3   | 1.519 (10) | C22—H22C | 0.9800     |
| C2—H2A  | 0.9900     | C23—C25  | 1.481 (14) |

## supplementary materials

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|            |            |               |            |
|------------|------------|---------------|------------|
| C2—H2B     | 0.9900     | C24—C26       | 1.385 (11) |
| C3—C4      | 1.514 (10) | C24—C30       | 1.400 (11) |
| C3—C8      | 1.529 (10) | C25—H25A      | 0.9800     |
| C3—C7      | 1.548 (11) | C25—H25B      | 0.9800     |
| C4—C5      | 1.504 (9)  | C25—H25C      | 0.9800     |
| C4—H4A     | 0.9900     | C26—C27       | 1.398 (11) |
| C4—H4B     | 0.9900     | C27—C28       | 1.374 (13) |
| C5—C6      | 1.471 (10) | C27—H27       | 0.9500     |
| C6—C9      | 1.451 (9)  | C28—C29       | 1.381 (12) |
| C7—H7A     | 0.9800     | C29—C30       | 1.379 (11) |
| C7—H7B     | 0.9800     | C29—H29       | 0.9500     |
| C7—H7C     | 0.9800     | C30—H30       | 0.9500     |
| C8—H8A     | 0.9800     |               |            |
| C9—S1—C11  | 101.5 (3)  | C15—C14—C13   | 110.5 (5)  |
| C15—O2—C11 | 108.4 (5)  | O5—C14—H14    | 110.6      |
| C17—O3—C16 | 116.2 (6)  | C15—C14—H14   | 110.6      |
| C19—O5—C14 | 118.9 (6)  | C13—C14—H14   | 110.6      |
| C21—O7—C13 | 117.6 (6)  | O2—C15—C16    | 109.1 (6)  |
| C23—O9—C12 | 116.7 (7)  | O2—C15—C14    | 109.0 (5)  |
| C1—N1—C24  | 127.0 (6)  | C16—C15—C14   | 109.1 (5)  |
| C1—N1—H1   | 111 (6)    | O2—C15—H15    | 109.9      |
| C24—N1—H1  | 122 (6)    | C16—C15—H15   | 109.9      |
| N1—C1—C6   | 123.1 (6)  | C14—C15—H15   | 109.9      |
| N1—C1—C2   | 114.9 (6)  | O3—C16—C15    | 107.4 (6)  |
| C6—C1—C2   | 121.9 (6)  | O3—C16—H16A   | 110.2      |
| C3—C2—C1   | 116.7 (6)  | C15—C16—H16A  | 110.2      |
| C3—C2—H2A  | 108.1      | O3—C16—H16B   | 110.2      |
| C1—C2—H2A  | 108.1      | C15—C16—H16B  | 110.2      |
| C3—C2—H2B  | 108.1      | H16A—C16—H16B | 108.5      |
| C1—C2—H2B  | 108.1      | O4—C17—O3     | 123.4 (7)  |
| H2A—C2—H2B | 107.3      | O4—C17—C18    | 126.5 (8)  |
| C4—C3—C2   | 106.8 (6)  | O3—C17—C18    | 110.1 (7)  |
| C4—C3—C8   | 111.3 (6)  | C17—C18—H18A  | 109.5      |
| C2—C3—C8   | 108.8 (6)  | C17—C18—H18B  | 109.5      |
| C4—C3—C7   | 111.6 (7)  | H18A—C18—H18B | 109.5      |
| C2—C3—C7   | 110.5 (6)  | C17—C18—H18C  | 109.5      |
| C8—C3—C7   | 107.8 (7)  | H18A—C18—H18C | 109.5      |
| C5—C4—C3   | 110.7 (6)  | H18B—C18—H18C | 109.5      |
| C5—C4—H4A  | 109.5      | O6—C19—O5     | 122.8 (7)  |
| C3—C4—H4A  | 109.5      | O6—C19—C20    | 125.5 (8)  |
| C5—C4—H4B  | 109.5      | O5—C19—C20    | 111.8 (8)  |
| C3—C4—H4B  | 109.5      | C19—C20—H20A  | 109.5      |
| H4A—C4—H4B | 108.1      | C19—C20—H20B  | 109.5      |
| O1—C5—C6   | 122.3 (6)  | H20A—C20—H20B | 109.5      |
| O1—C5—C4   | 120.0 (7)  | C19—C20—H20C  | 109.5      |
| C6—C5—C4   | 117.6 (6)  | H20A—C20—H20C | 109.5      |
| C1—C6—C9   | 124.0 (6)  | H20B—C20—H20C | 109.5      |
| C1—C6—C5   | 116.6 (6)  | O8—C21—O7     | 124.4 (7)  |
| C9—C6—C5   | 119.3 (6)  | O8—C21—C22    | 124.4 (7)  |



|              |            |                 |            |
|--------------|------------|-----------------|------------|
| C3—C7—H7A    | 109.5      | O7—C21—C22      | 111.0 (7)  |
| C3—C7—H7B    | 109.5      | C21—C22—H22A    | 109.5      |
| H7A—C7—H7B   | 109.5      | C21—C22—H22B    | 109.5      |
| C3—C7—H7C    | 109.5      | H22A—C22—H22B   | 109.5      |
| H7A—C7—H7C   | 109.5      | C21—C22—H22C    | 109.5      |
| H7B—C7—H7C   | 109.5      | H22A—C22—H22C   | 109.5      |
| C3—C8—H8A    | 109.5      | H22B—C22—H22C   | 109.5      |
| C3—C8—H8B    | 109.5      | O10—C23—O9      | 124.2 (9)  |
| H8A—C8—H8B   | 109.5      | O10—C23—C25     | 125.4 (9)  |
| C3—C8—H8C    | 109.5      | O9—C23—C25      | 110.3 (8)  |
| H8A—C8—H8C   | 109.5      | C26—C24—C30     | 118.1 (7)  |
| H8B—C8—H8C   | 109.5      | C26—C24—N1      | 122.3 (7)  |
| C6—C9—S2     | 124.9 (5)  | C30—C24—N1      | 119.4 (7)  |
| C6—C9—S1     | 116.8 (5)  | C23—C25—H25A    | 109.5      |
| S2—C9—S1     | 118.4 (4)  | C23—C25—H25B    | 109.5      |
| O2—C11—C12   | 106.4 (5)  | H25A—C25—H25B   | 109.5      |
| O2—C11—S1    | 111.1 (5)  | C23—C25—H25C    | 109.5      |
| C12—C11—S1   | 110.2 (5)  | H25A—C25—H25C   | 109.5      |
| O2—C11—H11   | 109.7      | H25B—C25—H25C   | 109.5      |
| C12—C11—H11  | 109.7      | C24—C26—C27     | 121.9 (7)  |
| S1—C11—H11   | 109.7      | C24—C26—Cl1     | 119.3 (6)  |
| O9—C12—C11   | 111.2 (6)  | C27—C26—Cl1     | 118.8 (6)  |
| O9—C12—C13   | 103.8 (5)  | C28—C27—C26     | 118.1 (8)  |
| C11—C12—C13  | 109.0 (6)  | C28—C27—H27     | 121.0      |
| O9—C12—H12   | 110.9      | C26—C27—H27     | 121.0      |
| C11—C12—H12  | 110.9      | C27—C28—C29     | 121.5 (7)  |
| C13—C12—H12  | 110.9      | C27—C28—Cl2     | 119.5 (7)  |
| O7—C13—C14   | 108.5 (5)  | C29—C28—Cl2     | 118.8 (7)  |
| O7—C13—C12   | 105.6 (5)  | C30—C29—C28     | 119.7 (7)  |
| C14—C13—C12  | 113.5 (5)  | C30—C29—H29     | 120.2      |
| O7—C13—H13   | 109.7      | C28—C29—H29     | 120.2      |
| C14—C13—H13  | 109.7      | C29—C30—C24     | 120.6 (7)  |
| C12—C13—H13  | 109.7      | C29—C30—H30     | 119.7      |
| O5—C14—C15   | 105.8 (6)  | C24—C30—H30     | 119.7      |
| O5—C14—C13   | 108.6 (6)  |                 |            |
| C24—N1—C1—C6 | 179.5 (7)  | O9—C12—C13—C14  | -165.2 (6) |
| C24—N1—C1—C2 | 2.5 (11)   | C11—C12—C13—C14 | -46.6 (8)  |
| N1—C1—C2—C3  | 170.9 (6)  | C19—O5—C14—C15  | -146.1 (6) |
| C6—C1—C2—C3  | -6.1 (10)  | C19—O5—C14—C13  | 95.3 (7)   |
| C1—C2—C3—C4  | 40.8 (8)   | O7—C13—C14—O5   | -85.2 (6)  |
| C1—C2—C3—C8  | 161.1 (7)  | C12—C13—C14—O5  | 157.9 (6)  |
| C1—C2—C3—C7  | -80.7 (8)  | O7—C13—C14—C15  | 159.2 (6)  |
| C2—C3—C4—C5  | -61.7 (8)  | C12—C13—C14—C15 | 42.2 (8)   |
| C8—C3—C4—C5  | 179.6 (7)  | C11—O2—C15—C16  | -170.2 (5) |
| C7—C3—C4—C5  | 59.1 (8)   | C11—O2—C15—C14  | 70.8 (7)   |
| C3—C4—C5—O1  | -132.3 (8) | O5—C14—C15—O2   | -170.0 (5) |
| C3—C4—C5—C6  | 50.7 (9)   | C13—C14—C15—O2  | -52.6 (8)  |
| N1—C1—C6—C9  | -0.5 (12)  | O5—C14—C15—C16  | 71.0 (7)   |
| C2—C1—C6—C9  | 176.3 (7)  | C13—C14—C15—C16 | -171.6 (6) |

## supplementary materials

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|                |            |                 |            |
|----------------|------------|-----------------|------------|
| N1—C1—C6—C5    | 174.0 (7)  | C17—O3—C16—C15  | -127.5 (7) |
| C2—C1—C6—C5    | -9.1 (10)  | O2—C15—C16—O3   | 61.0 (7)   |
| O1—C5—C6—C1    | 169.7 (8)  | C14—C15—C16—O3  | -180.0 (6) |
| C4—C5—C6—C1    | -13.3 (10) | C16—O3—C17—O4   | -4.9 (11)  |
| O1—C5—C6—C9    | -15.4 (12) | C16—O3—C17—C18  | 173.0 (6)  |
| C4—C5—C6—C9    | 161.5 (7)  | C14—O5—C19—O6   | 2.1 (12)   |
| C1—C6—C9—S2    | 8.6 (11)   | C14—O5—C19—C20  | -177.9 (7) |
| C5—C6—C9—S2    | -165.8 (6) | C13—O7—C21—O8   | 4.9 (11)   |
| C1—C6—C9—S1    | -170.6 (6) | C13—O7—C21—C22  | -169.7 (7) |
| C5—C6—C9—S1    | 15.0 (9)   | C12—O9—C23—O10  | -7.7 (11)  |
| C11—S1—C9—C6   | -175.6 (5) | C12—O9—C23—C25  | 169.7 (7)  |
| C11—S1—C9—S2   | 5.2 (5)    | C1—N1—C24—C26   | -92.1 (10) |
| C15—O2—C11—C12 | -75.1 (6)  | C1—N1—C24—C30   | 93.4 (10)  |
| C15—O2—C11—S1  | 164.9 (4)  | C30—C24—C26—C27 | -0.5 (12)  |
| C9—S1—C11—O2   | -74.5 (5)  | N1—C24—C26—C27  | -175.1 (7) |
| C9—S1—C11—C12  | 167.7 (5)  | C30—C24—C26—C11 | -178.4 (6) |
| C23—O9—C12—C11 | 104.5 (7)  | N1—C24—C26—C11  | 7.0 (10)   |
| C23—O9—C12—C13 | -138.5 (6) | C24—C26—C27—C28 | 2.4 (13)   |
| O2—C11—C12—O9  | 174.8 (5)  | C11—C26—C27—C28 | -179.6 (7) |
| S1—C11—C12—O9  | -64.6 (6)  | C26—C27—C28—C29 | -3.3 (13)  |
| O2—C11—C12—C13 | 61.0 (7)   | C26—C27—C28—C12 | -179.5 (7) |
| S1—C11—C12—C13 | -178.4 (5) | C27—C28—C29—C30 | 2.2 (14)   |
| C21—O7—C13—C14 | 95.4 (7)   | C12—C28—C29—C30 | 178.4 (7)  |
| C21—O7—C13—C12 | -142.7 (6) | C28—C29—C30—C24 | -0.1 (13)  |
| O9—C12—C13—O7  | 76.2 (6)   | C26—C24—C30—C29 | -0.7 (11)  |
| C11—C12—C13—O7 | -165.3 (5) | N1—C24—C30—C29  | 174.1 (7)  |

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

| $D-H\cdots A$     | $D-H$    | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-------------------|----------|-------------|-------------|---------------|
| N1—H1 $\cdots$ S2 | 0.88 (1) | 2.07 (5)    | 2.882 (6)   | 152 (9)       |

Fig. 1

